



## Supporting Information

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Nonsacrificial Nitrile Additive for Armoring High-Voltage  $\text{LiNi}_{0.83}\text{Co}_{0.07}\text{Mn}_{0.1}\text{O}_2$  Cathode with Reliable Electrode–Electrolyte Interface toward Durable Battery

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### **Nonsacrificial Nitrile Additive for Armoring High-voltage $\text{LiNi}_{0.83}\text{Co}_{0.07}\text{Mn}_{0.1}\text{O}_2$ Cathode with Reliable Electrode-electrolyte Interface toward Durable Battery**

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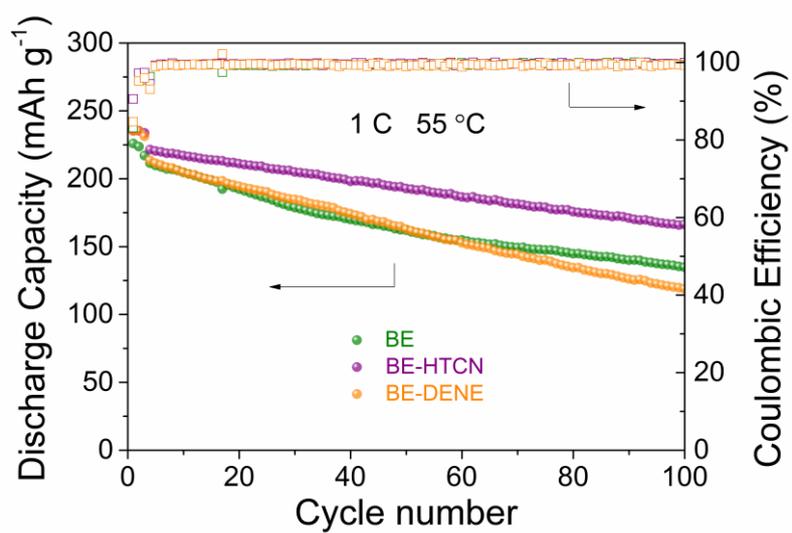
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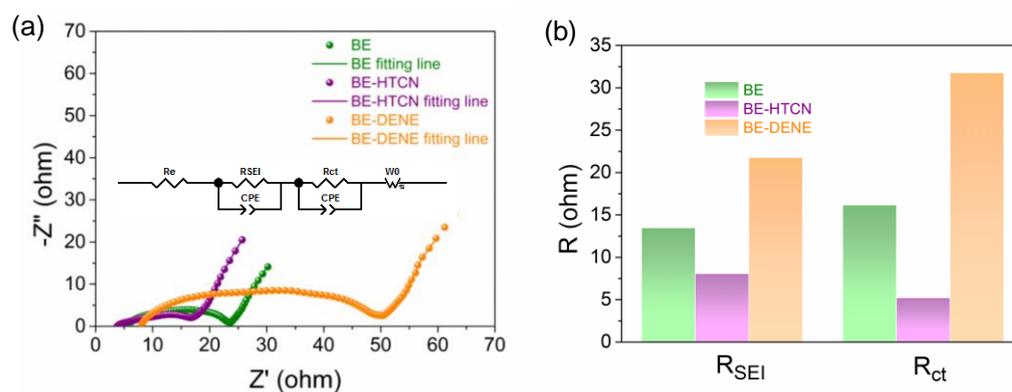
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**Table S1.** HOMO, LUMO energy of HTCN, DENE, EC, DEC and LiPF<sub>6</sub>.

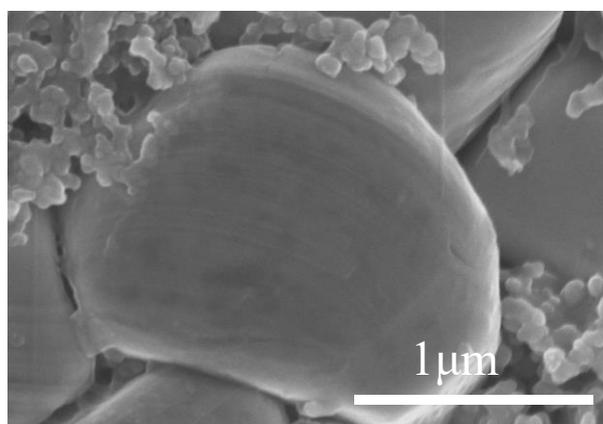
	HOMO	LUMO
HTCN	-8.050	-1.093
DENE	-6.563	-0.576
EC	-6.933	-0.300
DEC	-6.218	0.269
LiPF <sub>6</sub>	-8.851	-0.894



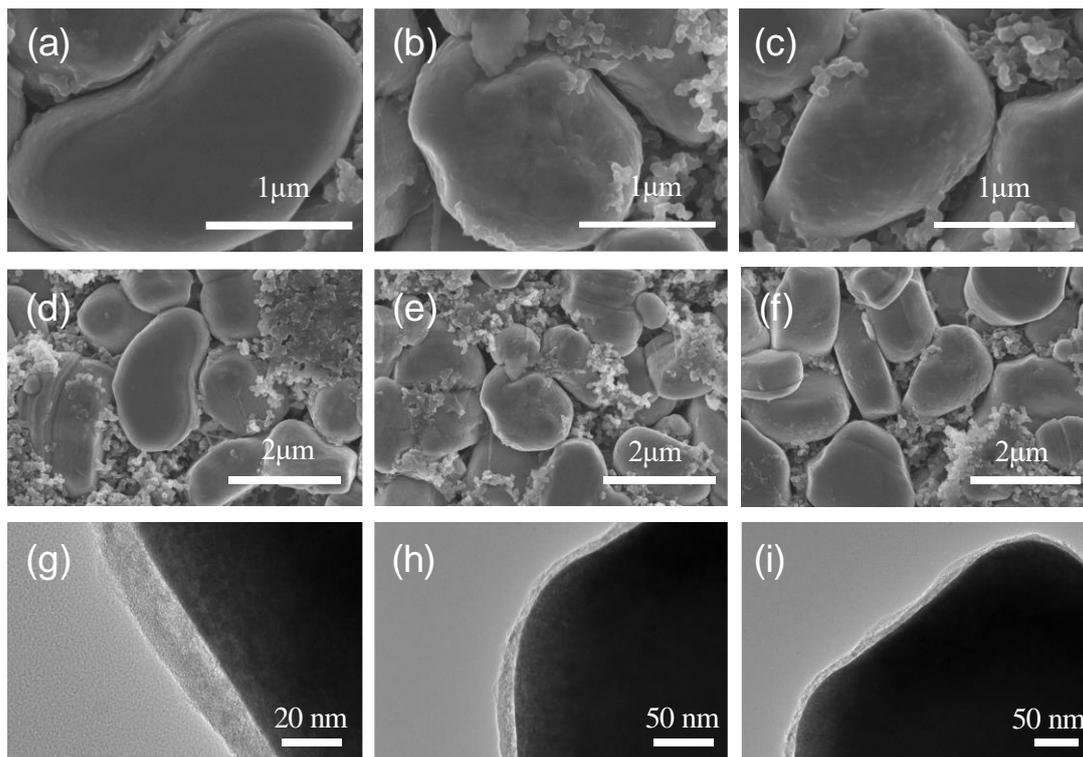
**Figure S1.** Cycle performance and coulombic efficiency of NCM83/graphite full cells with different electrolytes after 100 cycles at 55 °C.



**Figure S2.** (a) EIS Nyquist plots of NCM83/graphite full cells using BE, BE-HTCN and BE-DENE electrolytes (inset of equivalent circuit model). (b) The electrochemical impedance of the NCM83 electrode with three different electrolytes.



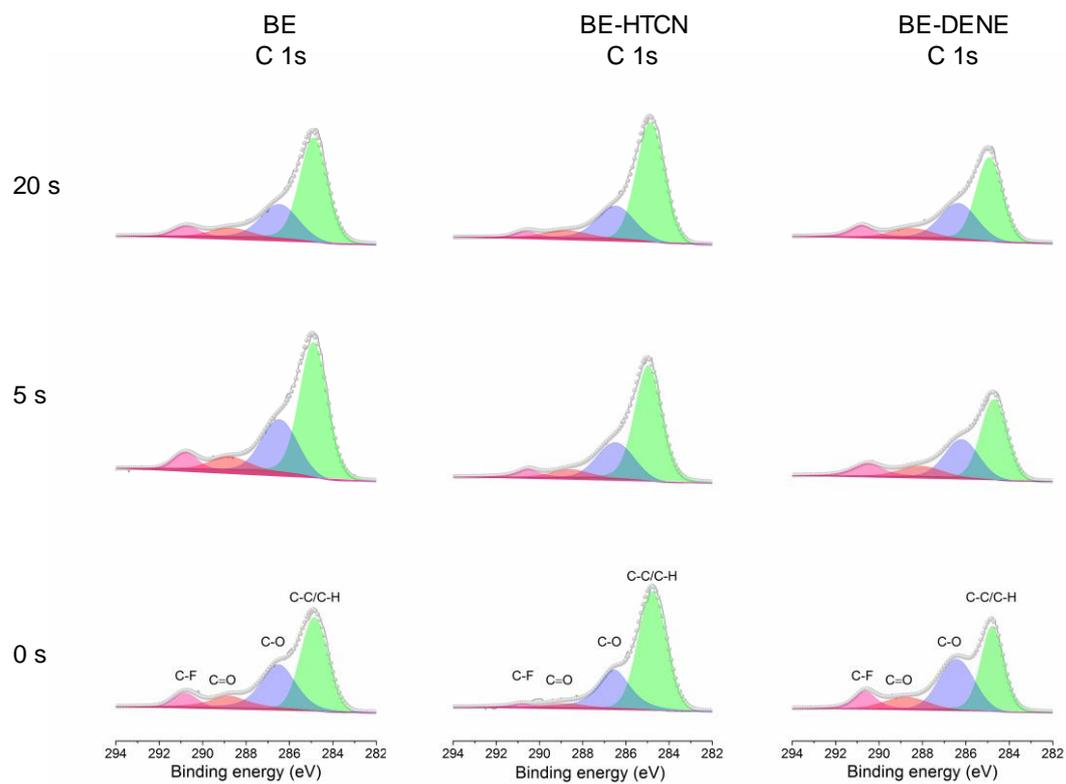
**Figure S3.** SEM image of the internal morphology of the particle before charge-discharge cycles.



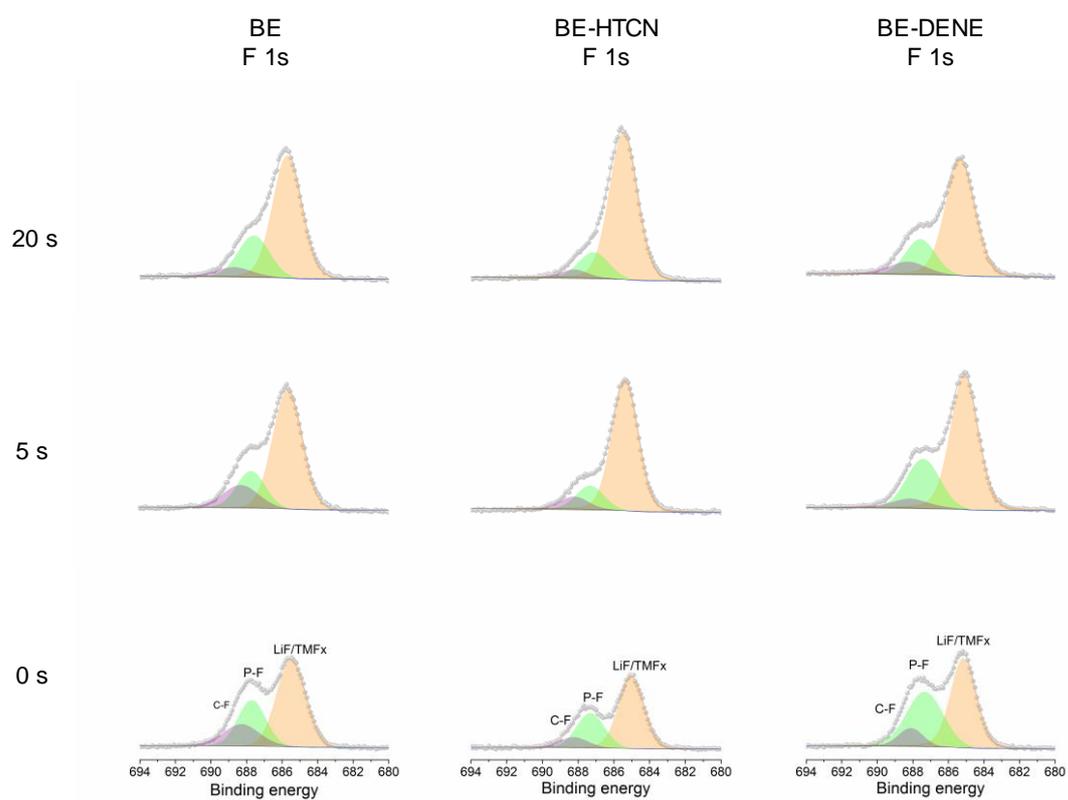
**Figure S4.** SEM and TEM images of the cycled NCM83 particles with BE-HTCN electrolyte. (a-f) SEM images, (g-i) TEM images.

**Table S2.** The surface contents of different elements on the cycled electrodes with the BE, BE-HTCN and BE-DENE electrolyte.

Elements	C 1s (%)	O 1s (%)	F 1s (%)	P 2p (%)	N 1s (%)
BE	20.27	17.68	45.32	0.54	---
BE-HTCN	21.72	22.72	34.06	0.33	5.46
BE-DENE	45.89	12.71	33.15	1.32	2.76



**Figure S5.** In-depth XPS spectra C 1s of CEI layer of NCM83/graphite cells in different electrolytes after 200 cycles for 0, 5 and 20 s, respectively.



**Figure S6.** In-depth XPS spectra F1s of CEI layer of NCM83/graphite cells in different electrolytes after 200 cycles for 0, 5 and 20 s, respectively.