



**PROGRAM** : MAIN EXAM FOR BENG DEGREE  
*EXTRACTION METALLURGY*

**SUBJECT** : FERROALLOY PRODUCTION

**CODE** : FAPMT 3B

**DATE** : 11 NOVEMBER 2019

**DURATION** : 3 HOURS

**WEIGHT** : 40 : 60

**TOTAL MARKS** : 100

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**EXAMINER** : DR X PAN

**MODERATOR** : K SEDUMEDI

**NUMBER OF PAGES** : 4 PAGES

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**INSTRUCTIONS** : ANSWER ALL QUESTIONS

**REQUIREMENTS** : CALCULATOR, RULER

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~~✓~~ 29/09/19  
Alli.  
03/10/2019

**QUESTION 1 ( 7 marks)**

Answer the following questions:

What is the chemical formula of chromite?

Give all names and chemical formulas of the 9 minerals that may exist in chrome ores

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**QUESTION 2 ( 13 marks)**

A diagram of O-C-Fe-Cr is used to help understand the smelting process of charge chrome, produced in a submerged arc furnace. Please draw the diagram with all names of the products.

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**QUESTION 3 ( 10 marks)**

Control of electrode penetration in a submerged arc furnace is one of the main challenges facing the production of charge chrome in South Africa. Please give 5 reasons and explain how they affect the penetration of an electrode.

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27/09/19

Belli  
Dr Kaelo Sedlemedi  
26/09/2019

**QUESTION 4 ( 20 marks)**

Outokumpu is one of the production processes used to produce charge chrome in South Africa. Please use I-P-O process model to answer the following questions of the process:

1. Give the details of 5 raw materials used as inputs
  2. Give the details of 3 main outputs
  3. Draw the process flow-sheet
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**QUESTION 5 (50 marks)**

(1) Using charge recipe calculation with the attached conditions, please answer the following questions:

1. SiO<sub>2</sub> wt-% in SiO<sub>2</sub>-MgO-Al<sub>2</sub>O<sub>3</sub> slag?
2. How many kg of quartz?
3. How many kg of coke?

(2) Using the 5-zone model of Dr Pan with the attached conditions, please calculate the mass balance in each zone

The required conditions are as the followings:

- **985 kg of ore**
- Final alloy with **3-5% Si, and 6-8% C**
- Final slag with **12-14%Cr<sub>2</sub>O<sub>3</sub>, and 6-8%FeO**
- The chemical compositions of raw materials are given in Table 1. The required liquidus temperature is 1700 °C for the slag of SiO<sub>2</sub>-MgO-Al<sub>2</sub>O<sub>3</sub>.
- The atomic weights of some elements are listed in Table 2, and the SiO<sub>2</sub>-MgO-Al<sub>2</sub>O<sub>3</sub> phase diagram is given in Figure 1.

Please submit the Figure 1 together with your answer sheet/s.

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23/09/19

Asli  
Dr Kaela Sedeemoli  
26/09/2019

**MAIN EXAM FOR FERROALLOY PRODUCTION**

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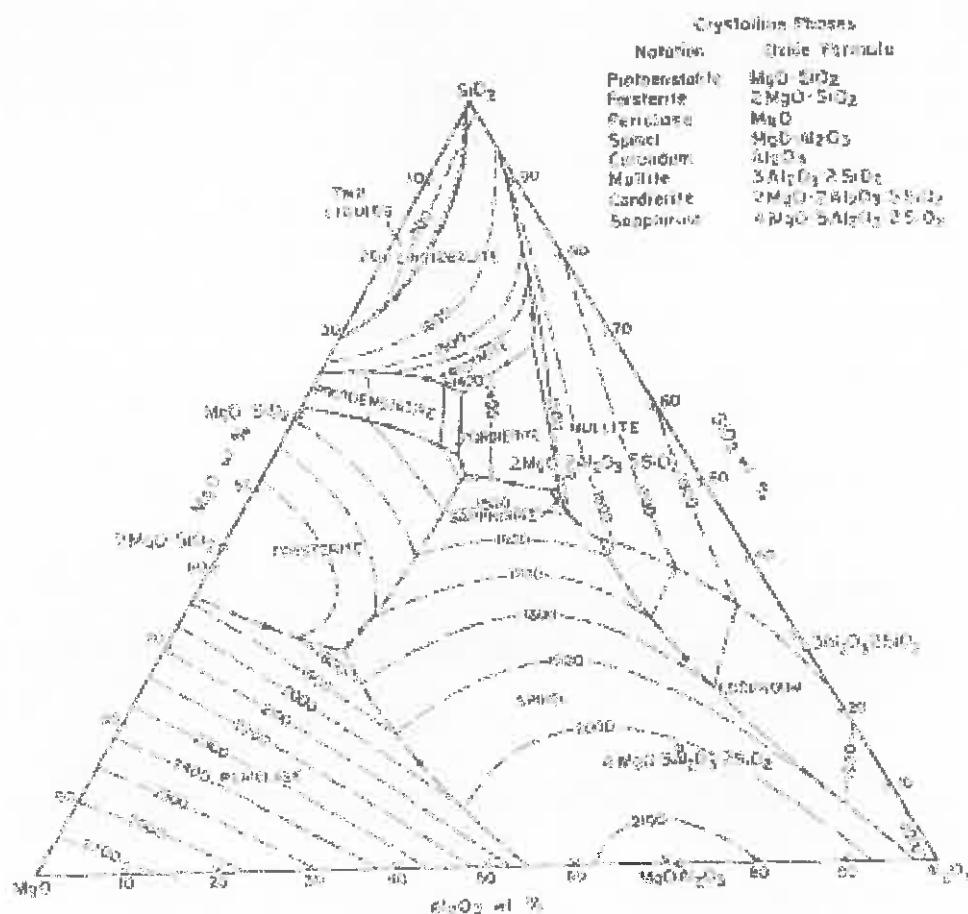
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**Table 1. Raw Material Composition**

Name	Cr <sub>2</sub> O <sub>3</sub> %	FeO%	Fe <sub>2</sub> O <sub>3</sub> %	MgO%	SiO <sub>2</sub> %	Al <sub>2</sub> O <sub>3</sub> %	H <sub>2</sub> O%	C%
Ore	39	23	0	12	9	12	5	0
Quartz	0	0		0	100	0	0	0
Coke	0	0		0	7	4	0	89

**Table 2. Atomic Weight**

Element	Fe	Cr	Si	Al	Mg	O	C	H
Weight	56	52	28	27	24	16	12	1



**Figure 1. SiO<sub>2</sub>-MgO-Al<sub>2</sub>O<sub>3</sub> Phase Diagram**

**Total = 100%**

*Don*  
23/09/19

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26/09/2019