



PROGRAM : *BACHELOR OF ENGINEERING TECHNOLOGY*
CHEMICAL ENGINEERING

SUBJECT : *CHEMICAL ENGINEERING FUNDAMENTALS*
CODE : *CEFCHB1*

DATE : 21 NOVEMBER 2017 (SUMMER EXAMINATION)

DURATION : 12:30 - 15:30 (SESSION2)

WEIGHT : 40: 60

TOTAL MARKS : 100
FULL MARKS : 100

EXAMINERS : MR ISHMAEL RAMATSA

MODERATOR : DR KAPIL MOOTHI

NUMBER OF PAGES : EIGHT (8) INCLUDING THIS COVER PAGE AND
APPENDICES

INSTRUCTIONS : ANSWER ALL QUESTION
: NUMBER ALL QUESTIONS IN ANY ORDER
: UNIVERSITY EXAM RULES APPLY

REQUIREMENTS : NON PROGRAMMABLE CALCULATOR

QUESTION 1 (30)

1.1 A 0.50-molar aqueous solution of sulfuric acid flows into a process unit at a rate of $1.25\text{m}^3/\text{min}$. The specific gravity of the solution is 1.03.

Calculate

1.1.1 The mass concentration of H_2SO_4 in kg/m^3 (3)

1.1.2 The mass flow rate of H_2SO_4 in kg/s . (3)

1.1.3 The mass fraction of H_2SO_4 . (5)

1.2 The heat capacity of ammonia, defined as the amount of heat required to raise the temperature of a unit mass of ammonia by precisely 1°F at a constant pressure, is, over a limited temperature range, given by the following expression.

$$C_p \left(\frac{\text{Btu}}{\text{lb}_m \cdot ^\circ\text{F}} \right) = 0.487 + 2.29 \times 10^{-4} T(^{\circ}\text{F})$$

1.2.1 Determine the expression for C_p in $\text{J}/(\text{g } ^\circ\text{C})$ in terms of $T(^{\circ}\text{C})$. (10)

1.3 Convert $23 \text{ lbm ft}/\text{min}^2$ to its equivalent in $\text{kg cm}/\text{s}^2$. (2)

1.4 Consider the interval from 20°F to 80°F , for the following

1.4.1 Calculate the equivalent temperatures in $^\circ\text{C}$ and the interval between them. (4)

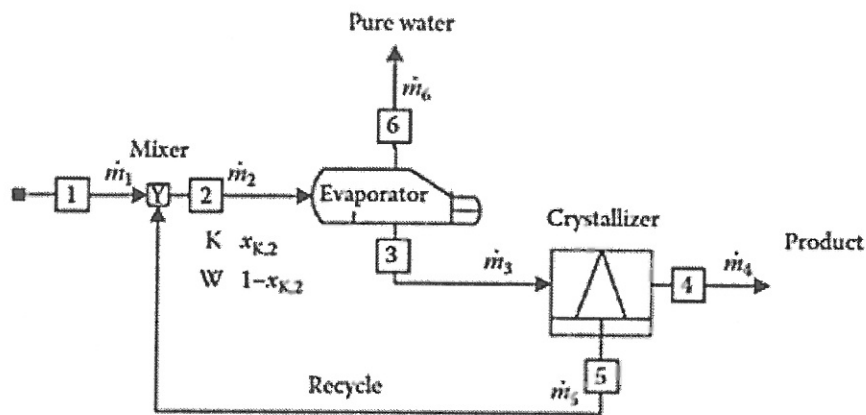
QUESTION 2 (30)

2.1 Fresh feed stream flowing at $27.77\text{g}/\text{s}$ contains 20% by weight KNO_3 (K) in H_2O (W). The fresh feed stream is combined with a recycle stream and is fed to an evaporator. The concentrated liquid solution exiting the evaporator contains 50% KNO_3 which is fed to a crystallizer. The crystals obtained from the crystallizer are 96% KNO_3 and the balance as water. The liquid from the crystallizer constitutes the recycle stream and contains 0.6 kg KNO_3 per 1000g of H_2O .

2.1.1 Calculate the degree of freedom around all the units and the process. (10)

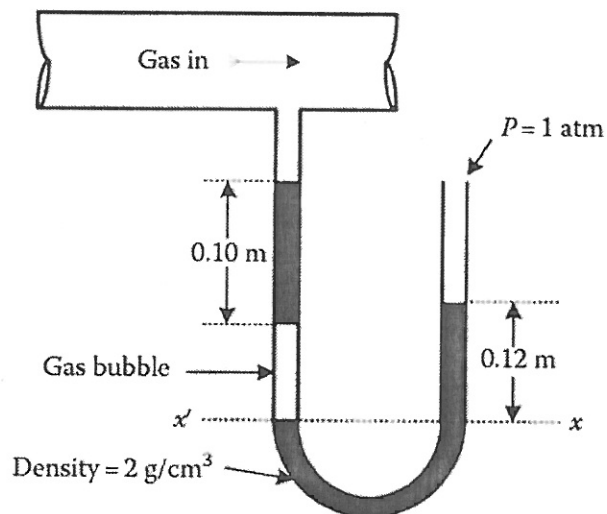
2.1.2 Re-draw the diagram and calculate all stream flow rate values in kg/h (12)

2.1.3 Calculate the compositions of water and the salt in stream 2 (8)



QUESTION 3 (15)

- 3.1 An open-end manometer is connected to a pipe in which a gas is flowing as shown (Figure below). The manometer fluid density is 2.00 g/cm^3 . A gas bubble is trapped in the left leg of the manometer (as shown). What is the pressure of the flowing gas inside the pipe (in atm)? Is this pressure a gauge pressure or an absolute pressure? How do you know? Assume that the density of the bubble is zero.



QUESTION 4 (25)

4.1 The feed to a compressor is superheated steam at 300°C and 20 bar absolute pressure. It enters the compressor at a velocity of 20 m/s. The pipe inlet inside diameter is 0.10 m. The discharging pipe, after the compressor, has a smaller inside diameter and the discharge velocity is 170 m/s. The exit of the compressor is superheated steam at 350°C and 60 bar absolute pressure. Heat loss from the compressor to the surroundings is 5 kW. Determine the compressor horsepower.
(15)

4.2 A closed rigid vessel that contains 200 kg of a fluid is heated from 20°C to 150°C. Calculate the heat required for this purpose. The constant volume heat capacity of the fluid is given by the following relation:
(10)

$$C_v \left(\frac{\text{kJ}}{\text{kg} \cdot ^\circ\text{C}} \right) = a + bT = 0.855 + 9.42 \times 10^{-4}T$$

Mass	$1 \text{ lb}_m = 5 \times 10^{-4} \text{ t} = 0.453593 \text{ kg} = 453.593 \text{ g} = 16 \text{ oz}$ $1 \text{ kg} = 1000 \text{ g} = 2.20462 \text{ lb}_m = 0.001 \text{ t}$ $1 \text{ t} = 2000 \text{ lb}_m; 1 \text{ t} = 1000 \text{ kg}$
Length	$1 \text{ ft} = 12 \text{ in.}; 1 \text{ ft} = 0.3048 \text{ m} = 30.48 \text{ cm}; 1 \text{ in.} = 2.54 \text{ cm}; 1 \text{ mile} = 5280 \text{ ft}$ $1 \text{ m} = 10^9 \text{ Å} = 39.37 \text{ in.} = 3.2808 \text{ ft} = 1.0936 \text{ yd} = 0.0006214 \text{ mi}$
Volume	$1 \text{ ft}^3 = 7.481 \text{ gal} = 1728 \text{ in.}^3 = 28.317 \text{ L} = 28.317 \text{ cm}^3$ $1 \text{ gal} = 231 \text{ in.}^3; 1 \text{ in.}^3 = 16.387 \text{ cm}^3$ $1 \text{ cc} = 1 \text{ cm}^3 = 1 \text{ mL}; 1000 \text{ mL} = 1 \text{ L}$ $1000 \text{ L} = 1 \text{ m}^3 = 35.3145 \text{ ft}^3 = 220.83 \text{ imperial gallons} = 264.17 \text{ gal} = 1056.68 \text{ qt}$ $8 \text{ fl oz} = 1 \text{ cup}; 4 \text{ cup} = 1 \text{ quart}; 4 \text{ quart} = 1 \text{ gal} = 128 \text{ fl oz}$
Density	$1 \text{ g/cm}^3 = 1 \text{ kg/L} = 1000 \text{ kg/m}^3 = 62.428 \text{ lb/ft}^3 = 8.345 \text{ lb}_m/\text{gal}$
Force	$1 \text{ lb}_f = 32.174 \text{ lb}_m \cdot \text{ft/s}^2 = 4.44822 \text{ N} = 4.4482 \times 10^5 \text{ dynes}$ $1 \text{ N} = 1 \text{ kg} \cdot \text{m/s}^2 = 10^5 \text{ dynes} = 10^5 \text{ g} \cdot \text{cm/s}^2 = 0.22481 \text{ lb}_f$
Pressure	$1 \text{ bar} = 10^5 \text{ Pa} = 100 \text{ kPa} = 10^5 \text{ N/m}^2$ Pascal (Pa) is defined as $1 \text{ N/m}^2 = 1 \text{ kg/m} \cdot \text{s}^2$ $1 \text{ atm} = 1.01325 \text{ bar} = 14.696 \text{ lb}_f/\text{in.}^2 = 760 \text{ mmHg at } 0^\circ\text{C (torr)} = 29.92 \text{ in Hg at } 0^\circ\text{C}$ $1 \text{ psi} = 1 \text{ lb}_f/\text{in.}^2; \text{psia (absolute)} = \text{psig (gauge)} + 14.696$
Temperature	$1 \text{ K} = 1.8^\circ\text{R (absolute temperature)}$ $T(^{\circ}\text{C}) = T(\text{K}) - 273.15$ $T(^{\circ}\text{F}) = T(^{\circ}\text{R}) - 459.67$ $T(^{\circ}\text{F}) = 1.8T(^{\circ}\text{C}) + 32$
Energy	$1 \text{ J} = 1 \text{ N} \cdot \text{m} = 1 \text{ kg} \cdot \text{m}^2/\text{s}^2 = 10^7 \text{ ergs} = 10^7 \text{ dyne} \cdot \text{cm} = 2.778 \times 10^{-7} \text{ kW} \cdot \text{h}$ $= 0.23901 \text{ cal} = 0.7376 \text{ ft} \cdot \text{lb}_f = 9.486 \times 10^{-4} \text{ Btu}$ $1 \text{ cal} = 4.1868 \text{ J}; 1 \text{ Btu} = 778.17 \text{ ft} \cdot \text{lb}_f = 252.0 \text{ cal}$ $1 \text{ Btu/lb}_m \cdot ^{\circ}\text{F} = 1 \text{ cal/g} \cdot ^{\circ}\text{C}$
Power	$1 \text{ hp} = 550 \text{ ft} \cdot \text{lb}_f/\text{s} = 0.74570 \text{ kW}$ $1 \text{ W} = 1 \text{ J/s} = 0.23901 \text{ cal/s} = 0.7376 \text{ ft} \cdot \text{lb}_f/\text{s} = 9.486 \times 10^{-4} \text{ Btu/s}$ $1 \text{ kW} = 1000 \text{ J/s} = 3412.1 \text{ Btu/h} = 1.341 \text{ hp}$

THE GAS CONSTANT (R)

$8.314 \text{ m}^3 \cdot \text{Pa}/(\text{mol} \cdot \text{K})$
$0.08314 \text{ L} \cdot \text{bar}/(\text{mol} \cdot \text{K})$
$0.08206 \text{ L} \cdot \text{atm}/(\text{mol} \cdot \text{K})$
$62.36 \text{ L} \cdot \text{mm Hg}/(\text{mol} \cdot \text{K})$
$0.7302 \text{ ft}^3 \cdot \text{atm}/(\text{lb-mole} \cdot ^{\circ}\text{R})$
$10.73 \text{ ft}^3 \cdot \text{psia}/(\text{lb-mole} \cdot ^{\circ}\text{R})$
$8.314 \text{ J}/(\text{mol} \cdot \text{K})$
$1.987 \text{ cal}/(\text{mol} \cdot \text{K})$
$1.987 \text{ Btu}/(\text{lb-mole} \cdot ^{\circ}\text{R})$

TABLE A.5 (Continued)

Properties of Superheated Steam

P (bar) (T_{sat} , °C)		Saturated Water	Saturated Steam	Temperature (°C)							
				50	75	100	150	200	250	300	350
40	<i>h</i>	1087.4	2830.3	212.7	317.1	422.0	634.3	853.4	1085.8	2962	3655
(120.3)	<i>v</i>	1082.4	2603.3	208.6	313.0	417.8	630.0	848.8	1080.8	2727	3629
	<i>u</i>	0.00125	0.04975	0.00101	0.00102	0.00104	0.00109	0.00115	0.00125	0.0088	0.0665
60	<i>h</i>	1213.7	2785.0	214.4	315.7	423.5	635.6	854.2	1085.8	2885	3646
(125.0)	<i>v</i>	1205.8	2590.4	208.3	312.6	417.3	629.1	847.3	1078.3	2668	3592
	<i>u</i>	0.00132	0.0325	0.00101	0.00103	0.00104	0.00109	0.00115	0.00125	0.0061	0.0422
80	<i>h</i>	1317.1	2759.9	216.1	313.3	425.0	636.8	855.1	1085.8	2787	3660
(126.0)	<i>v</i>	1306.0	2571.7	208.1	312.3	416.7	628.2	845.9	1075.8	2593	3550
	<i>u</i>	0.00139	0.0235	0.00101	0.00102	0.00104	0.00109	0.00115	0.00124	0.0245	0.0299
100	<i>h</i>	1408.0	2727.7	217.8	312.9	426.5	638.1	855.9	1085.8	1544	3626
(131.0)	<i>v</i>	1393.5	2547.3	207.8	311.7	416.1	627.3	844.4	1071.4	1329.4	3502
	<i>u</i>	0.00145	0.0161	0.00101	0.00102	0.00104	0.00109	0.00115	0.00124	0.0040	0.0224
150	<i>h</i>	1611.0	2615.0	222.1	316.0	430.5	641.3	858.1	1086.2	1336.2	3695
(142.0)	<i>v</i>	1586.1	2459.9	207.0	310.7	414.7	625.0	841.0	1067.7	1317.6	3523
	<i>u</i>	0.00166	0.0103	0.00101	0.00102	0.00104	0.00108	0.00114	0.00123	0.0038	0.0115
200	<i>h</i>	1826.5	2418.4	226.4	320.0	434.0	644.5	860.4	1086.7	1334.3	3647.1
(146.5)	<i>v</i>	1785.7	2300.8	206.3	309.7	413.2	621.9	837.7	1062.2	1307.1	3613.7
	<i>u</i>	0.00204	0.005875	0.00100	0.00102	0.00103	0.00108	0.00114	0.00122	0.0036	0.0067
221.20 (C)	<i>h</i>	2108	2136	228.2	331.7	435.7	645.8	861.4	1087.0	1332.8	3635.5
(174.05 K)	<i>v</i>	2037.8	2037.8	206.0	309.2	412.8	621.0	836.3	1060.0	1302.9	3601.3
	<i>u</i>	0.00317	0.00317	0.00103	0.00102	0.00103	0.00108	0.00114	0.00122	0.0035	0.0063
250	<i>h</i>	—	—	230.7	334.0	437.5	647.7	862.8	1087.5	1331.1	3650.0
(—)	<i>v</i>	—	—	205.7	308.7	412.1	620.8	834.4	1057.0	1297.5	3585.0
	<i>u</i>	—	—	0.00310	0.00301	0.00303	0.00308	0.00313	0.00322	0.00335	0.00360
300	<i>h</i>	—	—	235.0	338.1	441.6	650.9	865.2	1088.4	1328.7	3699.9
(—)	<i>v</i>	—	—	205.0	307.7	410.8	618.7	831.3	1052.1	1288.7	3563.3
	<i>u</i>	—	—	0.003480	0.00330	0.00333	0.00337	0.00343	0.00351	0.00363	0.00385
500	<i>h</i>	—	—	251.9	354.2	456.8	664.1	875.4	1093.6	1323.7	3763.3
(—)	<i>v</i>	—	—	202.4	304.0	405.8	611.0	819.7	1043.3	1259.3	3504.1
	<i>u</i>	—	—	0.003911	0.00360	0.00362	0.00366	0.00371	0.00379	0.00389	0.00404
1000	<i>h</i>	—	—	263.9	364.3	465.1	680.0	903.5	1115.0	1328.7	3850.5
(—)	<i>v</i>	—	—	196.5	295.7	393.1	594.4	795.3	999.0	1207.1	3419.0
	<i>u</i>	—	—	0.004337	0.003852	0.00380	0.00384	0.00388	0.00394	0.00402	0.00411
Temperature (°C)											
P (bar)		Saturated Water	Saturated Steam	400	450	500	550	600	650	700	750
0.0	<i>h</i>	—	—	3280	3384	3497	3607	3706	3816	3929	4043
	<i>v</i>	—	—	2669	3080	3132	3217	3303	3390	3480	3561
0.1	<i>h</i>	191.8	2544.8	3280	3384	3499	3606	3706	3816	3929	4043
(45.8)	<i>v</i>	191.8	2669	3080	3132	3217	3303	3390	3480	3561	3671
	<i>u</i>	0.00101	14.7	21.1	33.3	35.7	38.0	40.3	42.6	44.8	47.2
0.5	<i>h</i>	340.6	2646.0	3279	3383	3489	3596	3705	3816	3929	4043
(81.3)	<i>v</i>	340.6	2444.0	2969	3049	3132	3216	3302	3390	3480	3571
	<i>u</i>	0.00303	5.24	6.21	6.67	7.14	7.58	8.06	8.55	9.01	9.43
1.0	<i>h</i>	417.5	2675.4	3278	3382	3488	3596	3705	3816	3928	4042
(99.6)	<i>v</i>	417.5	2506.1	2968	3049	3132	3216	3302	3390	3479	3570
	<i>u</i>	0.00304	1.69	3.11	3.33	3.57	3.80	4.03	4.26	4.48	4.72

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