



PROGRAM : NATIONAL DIPLOMA  
ENGINEERING METALLURGY

SUBJECT : PRODUCTION OF IRON AND STEEL

CODE : PRS21-1

DATE : WINTER EXAMINATION 2016  
14 JUNE 2016

DURATION : (SESSION 2) 12:30 - 15:30

WEIGHT : 40 : 60

TOTAL MARKS : 100

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MODERATOR : MS N NAUDE 5063

NUMBER OF PAGES : 4 PAGES

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INSTRUCTIONS : ANSWER ALL QUESTIONS

REQUIREMENTS : CALCULATOR

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**QUESTION 1 (40 MARKS)**

1.1 Briefly explain reasons for the following pertaining to the blast furnace.

- (a) Particle size distribution of the burden has to be optimum. (2)
- (b) Steam is sometimes injected into the furnace. (2)
- (c) Sinter a better charge is better than lump ore. (2)
- (d) The shape of the furnace stack is an inverted cone and the hearth has smaller diameter than the bosh. (2)
- (e) The blast furnace is served by three stoves. (2)

1.2 Table 1 below shows the composition of blast furnace slag in percentage:

CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	MgO	FeO/Fe <sub>3</sub> O <sub>4</sub>	P	S
46	35	11	8	1	0.04	1.5

- (a) Why is sulphur much higher than phosphorus in slag? (2)
- (b) Calculate the basicity of slag and comment on its suitability. (2)
- (c) What does 1% FeO/Fe<sub>3</sub>O<sub>4</sub> show and what is the effect on the quality of iron? (2)
- (d) What would you do to improve blast furnace conditions with respect to (c)? (2)
- (e) What are the main sources of alumina and sulphur? (2)

1.3 How are the following conditions of the blast furnace corrected?

- (a) Slag and hot metal temperatures are too low. (2)
- (b) Slag basicity is 0.8 and slag is viscous. (2)
- (c) Top gas analysis is 16% CO, 15% CO<sub>2</sub>, 3% H<sub>2</sub> and 10% H<sub>2</sub>O. (2)
- (d) Top gas temperature is 1000°C. (2)
- (e) The furnace is driving too fast. (2)

1.4 Comment whether the metallurgist is correct or not on the following about blast furnace:

- (a) Blending of ores is unnecessary since reduction will still occur in the furnace. (2)
- (b) If the furnace is too hot, wet ore can be charged to cool the furnace. (2)
- (c) Phosphorus content in iron can be lowered by control of blast furnace process. (2)
- (d) Coke oven gas can be suitable as tuyere injection for the blast furnace. (2)
- (e) Tuyere injections boost productivity of the blast furnace. (2)

**QUESTION 2 (40 MARKS)**

2.1 What are the reasons for the following pertaining to the LD steel making process:

- (a) Mixers are necessary for hot metal storage prior to LD blowing. (2)
- (b) A desulphurising plant may not be necessary if blast furnace process is efficient. (2)
- (c) High phosphorus iron may be tolerated in LD but not high sulphur. (2)
- (d) High manganese iron may be tolerated in LD but not high silicon. (2)
- (e) High coke iron may be tolerated in LD but not high blast furnace slag. (2)

2.2 With reference to the blowing process briefly explain why:

- (a) Manganese and carbon are oxidized after silicon? (2)
- (b) Lime is charged in the early stages of the blow? (2)
- (c) The blow can sometimes be shortened to 15 minutes instead of 20? (2)
- (d) Phosphorus is oxidized right at the end of the oxygen blow? (2)
- (e) Carbon cannot be oxidised completely like silicon? (2)

2.3 With respect to end of blow i.e. "turn down", what action is taken if?

- (a) Temperature is too low e.g. 1599°C. (2)
- (b) Phosphorus is too high, e.g. 0.08%. (2)
- (c) Sulphur is too high, e.g. 0.09%. (2)
- (d) Parts of refractory lining are getting worn. (2)
- (e) There has been severe slopping during the blow. (2)

2.4 A 60 ton LD vessel has turn down analysis of 0.05% C, 0.15% Mn. Given that coke is 88% C, ferromanganese contains 80% Mn and ferrosilicon contains 64% Si, and that the customer requires a steel grade with 0.45% C and 0.85% Mn and 0.2% Si, calculate:

- (a) The amount of coke required. (2)
- (b) The amount of ferromanganese required. (2)
- (c) The amount of ferrosilicon required. (2)
- (d) What may cause the amounts calculated in (a), (b) and (c) to be inadequate. (2)
- (e) In view of your answer in (d), what steps will you take to always meet the grade? (2)

**QUESTION 3 (20 MARKS)**

3.1 Pertaining to iron making and steel making, explain why:

- (a) Iron making is a reduction process while steel making is an oxidising process. (2)
- (b) High  $\text{FeO/Fe}_2\text{O}_3$  is desirable in LD slag but not in blast furnace slag. (2)
- (c) Tapping temperatures are  $1660^\circ\text{C}$  for LD blown metal but  $1450^\circ\text{C}$  for hot metal. (2)
- (d) LD top gases are high in  $\text{O}_2$  while blast furnace top gases are high in  $\text{N}_2$ . (2)
- (e) Refractories for the LD basic, but neutral for the BF hearth? (2)

3.2 In comparison between blast furnace (BF) and LD steel making processes:

- (a) Write balanced equations for the main reactions providing heat for processes. (2)
- (b) BF and LD processes complement to control S and P in steel, why? (2)
- (c) BF and LD batch are processes, why is tap-to-tap time longer at 8 hours for BF? (2)
- (d) How does mineral dressing save on fluxes in BF? (2)
- (e) Why are the Boudouard and “carbon boil” the main reactions for the BF and LD. (2)

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Total = 100

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